What is Gravity?

- Gravity is a force that occurs between all objects
- Gravity always acts to pull objects towards each other
- The bigger the mass of an object, the more gravity it has



The sum effect of more than one force is called the resultant force.

The resultant force is calculated by working out the difference between opposing forces in each direction. What is the resultant force on this truck?

A resultant force of 100 N is accelerating the truck.





the Earth on the girl

Mass and Weight

- Mass is the amount of matter in an object and is measured in kilograms
- Weight is a force and is caused by the pull of gravity acting on a mass
- Weight is measured in newtons and has both magnitude and direction
- A newton meter is used to weight
- Gravity pulls the object downwards
- The amount the spring stretches tells us the force or weight

What is a force?

A force is a "push" or a "pull". Some common examples



Upthrust - keeps things afloat

Keywords

Equilibrium - State of an object when opposing forces are balanced

Deformation -Changing shape due to a force Linear relationship - When two variables are graphed and show a straight line which goes through the origin, and they can be called directly proportional.

Newton - Unit for measuring forces (N)

Resultant force - Single force which can replace all the forces acting on an object and have the same effect. **Friction** - Force opposing motion which is caused by the interaction of surfaces moving over one another. It is called 'drag' if one is a fluid.

Tension - Force extending or pulling apart

Compression - Force squashing or pushing together **Contact force** - One that acts by direct contact (friction) **Non Contact force** – Exerted without touching (gravity) **Range** -The maximum and minimum values of a variable. **Interval** -The gap between the values of the independent variable.

Control group - Those that are not exposed to the factor being tested.

Repeatable - When repeat readings are close together Balanced Force - Forces acting on a object are equal and opposite (no resultant force)

Streamlined - Shaped so that the flow of air around the body is made as smooth as possible

Weight vs. Mass

Earth's Gravitational Field Strength is 10N/kg. In other words, a 1kg mass is pulled downwards by a force of 10N.

Weight = Mass x Gravitational Field Strength (in ka) (in N/kg) (in N)



Moments

downwards

A moment is a "turning force", e.g. trying to open or close a door or using a spanner. The size of the moment is given by:

Moment (in Nm) = force (in N) x distance from pivot (in m)



Rule 1 - If the object is still, or moving at a constant speed the forces are balanced. Rule 2 - If the object is **speeding up** or **slowing down**, the forces are unbalanced.









- **Power** how quickly energy is transferred by a device (watts).
- Energy resource something with stored energy that can be released in a useful way
- Non-renewable an energy resource that cannot be replaced and will be used up.
- Renewable an energy resource that can be replaced and will not run out. Examples are solar, wind, waves, geothermal and biomass.
- Fossil fuels non- renewable energy resources formed from the remains of ancient plants or animals. Examples are, coal, crude oil and natural gas.







- Thermal energy store: Filled when an object is warmed up.
- **Chemical energy store:** Emptied during chemical reactions when energy is transferred to surroundings.
- Kinetic energy store: Filled when an object speeds up.
- Gravitational potential energy store: Filled when an object is raised.
- Elastic energy store: Filled when a material is stretched or compressed.
- **Dissipated:** Become spread out wastefully.





Typical value per 100 g	Per 30 g serving with 125 ml of semi-skimmed milk							
ENERGY	1639 kJ 387 kcal	743 kJ 175 kcal						
PROTEIN	5 g	6 g						
CARBOHYDRATE	85 a	32 g						
of which sugars	35 g	17 g						
starch	50 g	15 g						
FAT	2.5 g	3 g						
of which saturates	1 g	1.5 g						
FIBRE	2 g	0.6 g						
SODIUM	0.3 g	0.15 g						
SALT	0.75 g	0.35 g						
VITAMINS:	(% RDA)	(% RDA)						
VITABAINI D	1 2 110 (92)	12 10 (25)						

- When food burns, it releases chemical energy
- If the food is burned under a test tube containing water, chemical energy in the water is transferred to heat energy and the water heats up
- The hotter the water gets, the more energy there is in the food

Types of energy	Sources
Heat or thermal energy	Hot objects, e.g. fires
Light energy	The Sun, light, bulbs, fires, etc.
Sound energy	Vibrating objects e.g., Loudspeakers
Electrical energy	Available every time a current flows
Chemical energy	Food, fuels and batteries from chemical reactions.
Kinetic energy (the energy an object has because it is moving)	Flowing water, wind, etc.
Elastic potential energy	Objects such as springs and rubber bands that are stretched or twisted or bent
Gravitational potential energy	Objects that have a high position and are able to fall
Nuclear energy	Changes in the nucleus of certain heavy atoms e.g. Uranium.



- **Potential difference (voltage)**: The amount of energy shifted from the battery to the moving charge, or from the charge to circuit components, in volts (V).
- **Resistance**: A property of a component, making it difficult for charge to pass through, in ohms (Ω).
- **Current**: Flow of electric charge, in amperes (A).

voltage

current X resistance

Less resistance

More resistance

_ _ _ _ _ _ _ _ _ _ _ _ _

- Electrical conductor: A material that allows current to flow through it easily, and has a low resistance.
- Electrical insulator: A material that does not allow current to flow easily, and has a high resistance.

Electrons are tiny particles that carry a negative charge If an object:

- Gains electrons it becomes negatively charged
- Loses electrons it becomes **positively** charged



like charges repel



ROCK CYCLE IGNEOUS ROCK SEDIMENT

3 Major Rock Types

Igneous

- Formed from the solidification of molten rock (magmaor lava).
- Sedimentary
 - Formed at the Earth's surface from the accumulation and cementation of fragmented pieces of older rock produced by weathering.
- Metamorphic
 - Rocks that have undergone physical changes as a result of exposure to extreme pressure, temperature and fluids.



HOW IS A FOSSIL FORMED?



1. Sediment sediment, such as volcanic ash or silt, shortly after it dies. Its oones are protected rom rotting by the ayer of sediment.



2. Layers accumulate above the minerals, such as silica (a compound of silicon replace the calcium phosphate in





Movement 4. Erosion plates, or giant rock slabs that make up away the remaining Earth's surface, lifts rock layers. Eventually up the sediments and erosion or people digging for fossils will pushes the fossil expose the preserved closer to the surface







Earth and universe



in a year: 9.5 trillion km light year









Vibration: A back and forth motion that repeats.

Longitudinal wave: Where the direction of vibration is the same as that of the wave.

Transverse wave: Where the direction of the vibration is perpendicular to the direction of the wave.

Volume: How loud or quiet a sound is, in decibels (dB).

Pitch: How low or high a sound is. A low (high) pitch sound has a low (high) frequency.

Amplitude: The maximum amount of vibration, measured from the middle position of the wave, in metres.

Wavelength: Distance between two corresponding points on a wave, in metres.

Frequency: The number of waves produced in one second, in hertz.

Vacuum: A space with no particles of matter in it.

Oscilloscope: Device for viewing patterns of sound waves that have been turned into electrical current.

Absorption: When energy is transferred from sound to material.

Auditory range: The lowest and highest frequencies that a type of animal can hear.

Echo: Reflection of sound waves from a surface back to a listener.



mirror

Light: Electromagnetic radiation that can be detected by the human eye.

Ray: A narrow beam of light.

Medium: The substance the wave is travelling through, this could be a solid, liquid or gas.

Reflection: When light or any type of wave hits a new surface and returns in the direction it originated.

Refraction: The change in the direction of a wave when it passes from one medium into another.

Normal: A line drawn at 90° to the surface the ray of light is hitting.

Spectrum: The range of colours produced when white light passes through a prism.

Filter: Only allows certain wavelengths (colours) of light through, absorbing all others.

Transparent: A material that allows all light to pass through it.

Translucent: A material that allows most light through but not enough to make out detailed shapes.

Opaque: An object that lets no light through.

Dispersion: The splitting of white light into different wavelengths (colours)

Electromagnetic spectrum: range of frequencies of electromagnetic radiation and their respective wavelengths and photon energies.



- Work done = force x distance $W = F \times d$
- W is measures in joules.
- F is measure in newtons.
- **d** is measured in metres.
- When work is done, energy is transferred

Conduction



Convection





Key words

- Thermal conductor material that allows heat to move through it
- **Thermal insulator** material that only allows heat to move through it slowly
- **Temperature** a measure of particle movement
- Thermal energy amount of energy stored in a material due to particles vibrating
- **Conduction** transfer of thermal energy by the vibration of particles
- **Convection** transfer of thermal energy by particles rising
- **Radiation** transfer of thermal energy as a wave

Radiation



Speed – how much distance is covered on how much time. Average speed – overall distance travelled divided by overall time. Acceleration – how quickly speed increases. Fluid – no fixed shape, gas or liquid.

Pressure – ratio of force to surface area in N/m²





The magnetic field is strongest at the poles, where the field lines are most concentrated. Opposite poles attract, and like poles repel. Iron, Nickel and Cobalt are magnetic metals



When an electric current flows in a wire, it creates a magnetic field around the wire. This effect can be used to make an electromagnet.

We can make an electromagnet stronger by:

- wrapping the coil around an iron core
- adding more turns to the coil
- increasing the current flowing through the coil.

Electromagnets have some advantages over permanent magnets:

- they can be turned on and off
- the strength and direction of the magnetic field can be varied





Power station – NB: You need to understand the principle behind generating electricity. An energy resource is burnt to make steam to drive a turbine which drives the generator.

Energy

resource

gas)

Nuclear

Biofuel

Tides

Waves

Wind

Solar









		Radius of an atom 1 X 10 ⁻¹⁰ m						Decay	Rar	nge in air	Ionisin powe	Penetration po	ower			
Atom	Atom Same number of protons and electrons				//	Nucleus Decays to Another Nucleus		Alpha	Alpha Few cm		Very stro	ong Stopped by pa		X 🔶 - 3 o		
lon	Unequal number of electrons to protons		otons				Beta	F	Few m	Mediu	m Stopped by Alum	ninium Y	_β Υ *****			
Mass numb	Mass number Number of protons and neutrons		Parent Nucleus Daughter Nucleus		Gamma	Great	t distance	s Weak	Stopped by thic	ck lead Paper Al			uminium Lead			
Atomic nun	number Number of protons				lioactive	ble atoms randomly emit ation to become stable		-	Decay	Emitted from nuclei	Changes in mass eus number and atom					
Particle	Charg	Charge Size Found] \	deca	ay ecting		e Geiger I			Decay			numk		
Neutron	None	2 1			Unit			Becqu			Alpha (α)	Helium nuclei (⁴ ₂ He	;) -	4	-2	$^{238}_{92}U \rightarrow ~^{234}_{90}Th + ~^{4}_{2}He$
Proton	+	1	In the nucleus		Ionisation		A	All radiation ionises		s	Beta (β)	Electron $\begin{pmatrix} 0\\ -1 \end{pmatrix}$	(0	+1	$ \frac{{}^{14}_{6}C \rightarrow {}^{14}_{7}N + {}^{0}_{-1}e}{}^{0} $
Electron	-	Tiny	Orbits the nucleus	stru							Gamma (γ)	Electromagnetic wa	ve (0	0	$- \begin{array}{c} 99\\ 43\\ Tc \rightarrow \begin{array}{c} 99\\ 43\\ Tc + \end{array} \gamma$
	$\frac{1}{1}$ $\frac{1}$					Atoms and Isotopes			is and		Neutron	Neutron	-1	1	0	000- 901 1729- 1001 101- 101- 101- 101- 101- 101- 101
Isotope									clear	C	ontamination	Unwanted pres	esence of radioactive atoms			
Different forms of an element with the same					15	solopes	Radiatio		Ir	adiation	Person is in exp	osed to ra	idioacti	ive source	0 1 2 3 4 5	
number of	proton	s but different	number of neutrons			A	QA		Haza	rds an	d uses of				lose half	
	Di		ATOMIC Radioacti					S	of its initial radioactivity							
Democritus	Suggested idea of atoms as small spheres that cannot						STRUCTURE			of bac	kground	Sievert		Unit measuring dose of radiation		
	be cut.					(Separates) radiation Backgrou						Background	nd <i>Constant low level environmental radiation,</i> <i>e.g. from nuclear testing, nuclear power,</i>			
J JDiscovered electrons- emitted from surface of hotThomsonmetal. Showed electrons are negatively charged and						/	,								wast	te
(1897)										Different isotopes have different half lives		Short half-lives used in high doses, long half lives used in low doses. Isotope with short half life injected, allowed to circulate and collect in damaged areas. PET scanner used to detect emitting radiation. Must be beta or gamma as alpha does not penetrate the body.				
Thomson (1904)		Proposed 'plum pudding' model – atoms are a ball of positive charge with negative electrons embedded in it.														
Geiger and	Directed beam of alpha particles (He ²⁺)at a thin sheet					n	Tracer	S Used with		thin body						
Marsden	en of gold foil. Found some travelled through, some were								ion U	Ised to tr	at illnesses	Cancer cells killed by gamma rays. High dose used to				-
(1909)			lence to suggest alpha				therapy			ancer	Damage to healthy cells prevented by focussed gamma ray gun.					
Butherford	lected due to ele		and	Fuel ro	ods	Made of U-238, 'enriched' with U-235 (3%). Long and thin to a						utrons to e	scape, hitting nuclei.			
Rutherford (1911)very small charged nucleus, nucleus was massive.Proposed mass and positive charge contained in					Contro Concre				Made	e of Boron. Co	ontrols the rate of react	ion. Boror	Boron absorbs excess neutrons.			
	ucleus while eleo which cancel		liss	Concre	ete		Ne	Neutrons hazardous to humans – thick				< concreate shield protects workers.				
			rn model of atom – ele							Neu	tron hits U-23	5 nucleus, nucleus	Process	-		Compound nucleus
 circular orbits around nucleus, electrons can change orbits by emitting or absorbing electromagnetic radiation. His research led to the idea of some particles within the nucleus having positive charge; these were 					Nuclear Nuclear	One large unstable cleus splits to make wo smaller nuclei vo small nuclei join		absorbs neutron, splits		plits emitting two or I two smaller nuclei.		tion for				
					ZZŻ					eleases energy.	Used in s	nucleaı stations	-	Deuterium Hélium		
	named protons.							c Tw		ifficult to do	on Earth – huge				Fusion ++	
Chadwick	Discovered neutrons in nucleus – enabling other scientists to account for mass of atom.					Nuclear		_			ints of pressu	re and temperature	Occ	urs in stars		Energy
(1932) scientists to account for mass of atom. $\vec{z} \neq nucleus$ needed.											Tritium Neutron					





















A.			A natu A body ravity w	rge body orbiting the Sun ral satellite orbiting a planet large enough to have its own which caused a spherical shape oject orbiting the Sun due to gravity	comets, asteroids, sa	PLANETS DWARF PLANETS Durb at the	Effect of gravity.	 planets, stars Force o moon's c Gravity p 	f planets to to orbit ga f gravity ch		, [a	Too slow	steady = falls culate s	orbit around Earth. to Earth. speed of ce object
Milky Way our galaxy.		Galaxy Iniverse		lection of billions of stars Collection of galaxies The life cycle of a sta	Other objects.	tem			motio	of Or	bit.	Distar ave dist	rage sp ance ÷	∏r, then peed =
Nebula Protost	ar	A cloud of co hydrogen go and dust The large bo gas contract form a star	as all of ts to	Cloud collapses due to gravity, pa fast colliding with each other, kin into internal energy and the tem High temperature causes Hydrog and nuclear fusion begins. A star	articles move very etic energy transfers perature increases. en nuclei to collide		SP oara	QA PACE ates only shift		A planet's vertices of the second sec	elocity speed istant. un's ets	When ambul go past the s	Pla Pla Su ances	trong. Planets move quickly. anets further away from the in, gravity pull is weaker. So speed of planet is slower. Frequency of sound wave decreases,
Main sequen		Stable perio star	a of	Gravity tries to collapse the star l pressure of fusion energy expand inward force. ne size as our Sun.			7	Red-shift		the Sun and changes direc erved increase nost distance	tion.	changes from pitch to a low length of light	-	Galaxies are moving away
Red giant	fuse	rge star that s Helium int vier element	t Hyd dro res	drogen runs out, star becomes ur ps causing star to collapse. Atom ults in atoms fusing and tempera rease in temperature causes the	s now closer together ture increases. This	ding models.		Hubble (1929)	towo He studi	ards the red en ied light from ency decreases	nd of the distant g 5, wavele	-		from us in all directions. Light from distant galaxies is red-shifted, so galaxy is moving away
White dwarf Black		r collapses		clear fuel runs out, fusion stops,	dense very hot core.	Understand				_		earby galaxy. listant galaxy.		from us. Galaxies further away have bigger red-shift so
dwarf		Sta	Irs large	er than our Sun.		n		The Big Ban All matter an violently fro	nd space ex		ed—shift			are moving faster away.
Red sup giant	ber	Star swells greatly Gigantic	5	Nuclear fuel begins to run out matter = bigger size). Rapid collapse, heats to very h causing run away nuclear reac	ligh temperatures	(an	stotl stien	e It Greek)	Earth at	the centre, otl around	her heave the Earth	enly bodies mo h.		Planets and moons
Supernova		explosion (run away j reactions	on due to ay fusionflinging remnants out into space forces collapse the core into a		ce. Large gravitational tiny space.	(:	Copernicus (1473 - 1543) Galileo (1610)		Sun at the centre, other heavenly bodies move around the Sun. Made a telescope, looked at Jupiter, found four moons rotating around planet.			moved at different speeds to stars = reason for different positions.		
Neutro star	n	Very dense	e star	Made out of neutrons.]	

OR if collapse is into a really tiny space.

Black hole No light escapes Gravitat